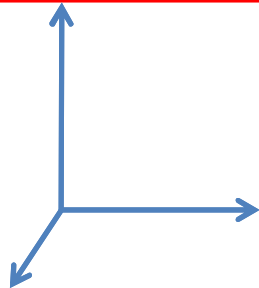


# Thermodynamics in 15 Minutes

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DMSE/PPST

$$dU = \sum_i \mathbf{F}_i d\mathbf{x}_i = \sum_i \left( \frac{\partial U}{\partial \mathbf{x}_i} \right)_{\mathbf{x}_{j \neq i}} d\mathbf{x}_i$$

There is a potential ENERGY (U) that represents the sum of system descriptors (Coordinates) times their respective partials with respect to energy (Forces).



$$\hat{\mathbf{x}}_i \cdot \hat{\mathbf{x}}_j = \delta_{ij}$$

Generally, all coordinate axis are orthogonal...

$-P, \gamma, \mathcal{H}, \mathcal{E}$   
 $V, A, \mathcal{M}, \mathcal{P}$

How do we fit heat into this equation?

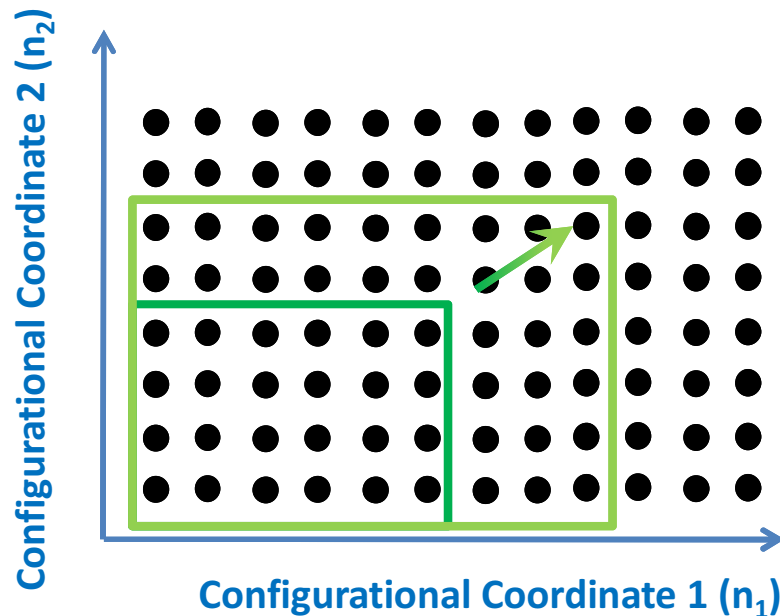
**S? T?**

Meet the new boss, same as the old boss...only in PHASE SPACE!

$$\delta W = -P dV = \sum_i \mathbf{F}_i d\mathbf{x}_i$$

$$\delta Q = \sum_i \mathbf{F}_{i,\text{conf}} d\mathbf{x}_{i,\text{conf}}$$

Analogy between  
P-V and T-S



$$\mathbf{F}_{i,\text{conf}} \rightarrow \sim kT/n_i$$

$$d\mathbf{x}_{i,\text{conf}} \rightarrow dn_i$$

$$\delta Q_i = \frac{kT}{n_i} dn_i = Td(k \ln n_i) = TdS_i$$

\*Here we invoke equipartition – a bit circular, so NOT a rigorous proof!

## *The Hierarchy*

Complete Equation:

$$dU = \delta Q + \delta W = \left( \frac{\partial U}{\partial S} \right)_{V, N, \dots} dS + \left( \frac{\partial U}{\partial V} \right)_{S, N, \dots} dV + \left( \frac{\partial U}{\partial N} \right)_{V, S, \dots} dN + \dots$$

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Needs Inputs:

**T**

**-P**

**$\mu$**

$$\left( \frac{\partial U}{\partial S} \right)_{V,N,\dots} = \mathbf{T}(S, V, N, \dots)$$

Equations of State: First Partial Derivatives  
(But how do we determine these?)

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Which Need Parameters:

$$\left( \frac{\partial^2 U}{\partial S \partial V} \right)_{P,N,\dots} = - \frac{1}{V\alpha}$$

Material Constants: Second Partial Derivatives  
(Determined Experimentally or w/ Stat Mech)

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Relationships at each level lowers degrees of freedom:

$$SdT - VdP + Nd\mu = 0$$

$$\left( \frac{\partial}{\partial S} \left( \frac{\partial U}{\partial V} \right)_{S,N} \right)_{V,N} = \left( \frac{\partial}{\partial V} \left( \frac{\partial U}{\partial S} \right)_{V,N} \right)_{S,N}$$

## Addition of Constraints

How do we take this:

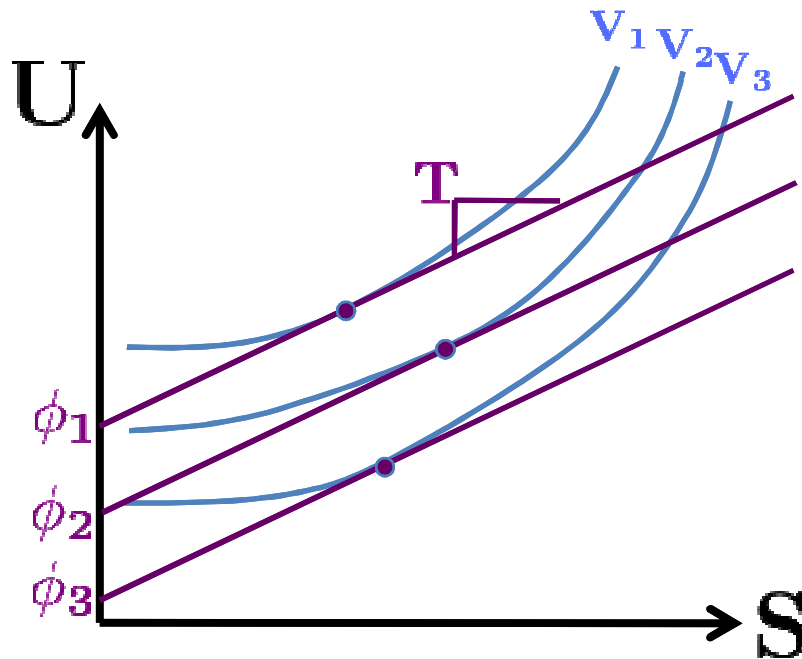
$$U(S, V, N) = TS - PV + \mu N$$

And incorporate this:

Or this:

$$T = \text{Constant}$$

$$P = \text{Constant}$$



Define a new potential:

$$\phi_i = U_i - TS_i$$

This is Helmholtz Free Energy!  
We can define others depending on specific constraints.

*The Microscopic Picture*

Previously we determined:

$$S = k \ln \mathbf{n} \rightarrow S_i = -Nk(P_i \ln P_i)$$

And we just derived for a constant T system:

$$F_i = U_i - TS_i$$

Combine these to get:

$$N_i(F_i - U_i) = kTN_i \ln P_i$$

$$P_i = e^{-U_i/kT} e^{F/kT} = \frac{e^{-U_i/kT}}{Q}$$

Since  $P_i$  must be normalized:

$$Q = \sum_i e^{-U_i/kT} = \sum_U n(U) e^{-U/kT}$$

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Replace summation by most likely value (\*) (large N)

$$Q \approx e^{\ln n^* - U^*/kT} = e^{-F^*/kT}$$

$$F^* = -kT \ln Q$$

Plug this BACK INTO THERMO EQUATIONS!

$$dF = -TdS - PdV$$

$$-kT \left( \frac{\partial \ln Q}{\partial V} \right)_S = -P$$

## Map of Thermo

### FUNDAMENTAL EQUATION

$$dU = \sum_i F_i dx_i \xrightarrow{\text{Constraints } j} \phi = U - \sum_j F_j x_j$$

